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Paradigm of engineering recalcitrant non-model microorganism with dominant metabolic pathway as a biorefinery chassis

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The development and implementation of microbial chassis cells have profound impacts on circular economy. Non-model bacterium Zymomonas mobilis is an excellent chassis owing to its extraordinary industrial characteristics. Here, the genome-scale metabolic model iZM516 is improved and updated by integrating enzyme constraints to simulate the dynamics of flux distribution and guide pathway design. We show that the innate dominant ethanol pathway of Z. mobilis restricts the titer and rate of these biochemicals. A dominant-metabolism compromised intermediate-chassis (DMCI) strategy is then developed through introducing low toxicity but cofactor imbalanced 2,3-butanediol pathway, and a recombinant D-lactate producer is constructed to produce more than 140.92 g/L and 104.6 g/L D-lactate (yield > 0.97 g/g) from glucose and corncob residue hydrolysate, respectively. Additionally, techno-economic analysis (TEA) and life cycle assessment (LCA) demonstrate the commercialization feasibility and greenhouse gas reduction capability of lignocellulosic D-lactate. This work thus establishes a paradigm for engineering recalcitrant microorganisms as biorefinery chassis.

The improvement on sustainable waste management and resource utilization efficiency can help reduce pollution and strengthen sustainable biochemical production to achieve the goal of circular economy¹. Non-food materials such as lignocellulose, glycerol, molasses, and waste gas serve as sustainable feedstocks of biomanufacturing²⁻⁴. The construction of efficient microbial cell factories is crucial for the sustainable conversion of low-cost non-food materials into high-value biochemicals⁵.

The rapid progress in systems biology and synthetic biology has greatly accelerated the development of efficient genome-engineering toolkits and effective microbial cell factories for diverse biochemicals production^{5,6}. However, most achievements in this field have been accomplished using model microorganisms such as *Escherichia coli*, *Saccharomyces cerevisiae*, *Corynebacterium glutamicum*, and *Bacillus subtilis* because of their clear genetic background, comprehensive genome-editing tools, as well as abundant biological parts and devices⁷⁻¹⁰.

Given the limited number of model species and inherent drawbacks associated with their use in industrial production, significant efforts have been focused on identifying and developing non-model microorganisms with excellent characteristics for biomanufacturing, such as *Vibrio natriegens* (fast growth)¹¹, *Acinetobacter baylyi* (high

¹State Key Laboratory of Biocatalysis and Enzyme Engineering, Environmental Microbial Technology Center of Hubei Province, and School of Life Sciences, Hubei University, Wuhan, China. ²Xi'an Key Laboratory of C1 Compound Bioconversion Technology, School of Chemical Engineering and Technology, Xi'an Jiaotong University, Xi'an, China. ³Biodesign Center, Key Laboratory of Systems Microbial Biotechnology, Tianjin Institute of Industrial Biotechnology, Chinese Academy of Sciences, Tianjin, China. ⁴These authors contributed equally: Xiongying Yan, Weiwei Bao. e-mail: feiqiang@xjtu.edu.cn; Qiaoninghe@hubu.edu.cn; Shihui.Yang@hubu.edu.cn natural transformation efficiency)¹², Lactococcus lactis (lactate production)¹³, and *Halomonas bluephagenesis* (PHA production)¹⁴. In addition, some non-model microorganisms possess diversified industrial characteristics that make them suitable for engineering as a biorefinery chassis, such as robustness against severe environmental conditions as well as toxic inhibitors, biosynthetic intermediates and end-products, broad spectrum of substrate utilization and biochemical production capabilities, high yield of target products with few byproducts, innate immunity from phage infection and stable genome structure¹⁵⁻¹⁷. However, compared to model microorganisms, the lack of abundant information and efficient genome-engineering toolkits for non-model microorganisms limits their development and applications despite the continuous development of numerous tools for non-model species with microbial cell factories being constructed at an unparalleled speed for industrial production. Efficient and effective toolkits and pipelines for enzyme selection, rational pathway design, and genome engineering are still in their infancy. Tremendous efforts are needed to expedite the development of diverse biorefinery chassis and their derivative cell factories for sustainable circular bioeconomy. More importantly, non-model microorganisms require more systematic research, including the utilization of multiple feedstocks and production of diverse products, to become the biorefinery chassis while being developed into specific cell factories.

Zymomonas mobilis is a non-model polyploid ethanologenic Gram-negative bacterium with many industrial merits and unique physiological properties¹⁸, which is the only known microorganism that can utilize the Entner-Doudoroff (ED) pathway under anaerobic conditions¹⁹ and possesses excellent characteristics such as high sugar uptake rate, high ethanol yield and ethanol tolerance²⁰. Significant efforts have been continuously made to achieve accurate genome sequence and annotation^{21,22}, as well as significant amounts of omics datasets^{21–23}, which have been used for the prediction and expansion of biological parts such as RBS and promoters^{24,25}. In addition, efficient genome-editing tools based on heterologous CRISPR-Cas12a, endogenous Type I-F CRISPR-Cas, and associated repair pathways, such as

microhomology-mediated end joining (MMEJ) have been developed²⁶⁻²⁸. Several microbial cell factories have been constructed for the production of various biochemicals, including acetaldehyde, D-lactate, L-alanine, L-serine, acetoin, 2,3-butanediol (2,3-BDO), isobutanol, L-malate, succinic acid, poly-3-hydroxybutyrate (PHB), farnesene, and ethylene²⁹⁻⁴⁰ (Fig. 1). Moreover, an improved genome-scale metabolic model (GEM) of *i*ZM516 for *Z. mobilis* was recently constructed, which contains the most reactions, metabolites, and genes among all other reported models. This model was used to design the metabolic pathways for the production of succinate and 1,4-buta-nediol (1,4-BDO)⁴¹.

Despite these accomplishments, the primary obstacle in making *Z*. *mobilis* a chassis for non-food feedstock biorefinery^{19,42} lies in circumventing its dominant ethanol production pathway from pyruvate comprised of efficient pyruvate decarboxylase (PDC) and alcohol dehydrogenases (ADHs). A central-carbon metabolism control-valve strategy has been explored to redirect the carbon flux to the 2,3-BDO, lactate, or isobutanol pathways with ca 65% overall yield from glucose⁴³. Similarly, efforts have been made to shift ethanol metabolism towards lactate production by replacing the strong promoter of *pdc* with an inducible promoter *Ptet*³¹. However, none of these studies have succeeded in completely eliminating *Z. mobilis'* dominant ethanol pathway, thereby limiting its potential as a biorefinery chassis for the production of biochemicals other than ethanol with a high titer, yield, and rate.

In this work, we improve the GEM *i*ZM516 of *Z. mobilis* ZM4 by integrating enzyme constraints to obtain ec*i*ZM547 for the rational design of metabolic pathways for C2-C5 biochemical production. We design and construct metabolic pathways for the production of 1,3-propanediol (1,3-PDO) from glycerol, butanediol from glucose, xylonic acid, ethylene glycol, glycolic acid, and 1,4-butanediol from xylose using *Z. mobilis* as the chassis. We then develop a metabolic strategy to bypass dominant metabolism by first constructing a metabolism-comprised intermediate chassis instead of directly engineering the chassis for the target biochemicals. Subsequently, we construct a



Fig. 1 | **Summary of using** *Z***.** *mobilis* **as the chassis for biochemical production.** The blue font in the pathway represents the key enzymes. The solid line represents a one-step reaction, while the dotted line indicates a multi-step reaction. The thick blue solid line represents the native recalcitrant ethanol pathway. 2,3-BDO 2,3butanediol, DMAPP dimethylallyl diphosphate, DXP 1-deoxy-D-xylulose-5-

phosphate, ED Entner-Doudoroff pathway, FPP farnesyl diphosphate, G3P glyceraldehyde-3-phosphate, G6P glucose-6-phosphate, GPP geranyl diphosphate, IPP isopentenyl diphosphate, MEP methylerythritol 4-phosphate pathway, PHB poly-3hydroxybutyrate, PPP pentose phosphate pathway, TCA tricarboxylic acid cycle.



Fig. 2 | **Comparisons of the maximum specific growth rates (h⁻¹) at different glucose uptake rates (mmol·gDW⁻¹·h⁻¹) under anaerobic condition using different models. a** Predictions for specific cell growth rates. **b** Predictions for ethanol production fluxes. *i*ZM4_478 is a model published recently⁴¹. *i*ZM516 is a revised

model version constructed in this study on the basis of the published one. eciZM547 is the enzyme-constrained version on the basis of revised *i*ZM516. The bullet points with different color and shape were experimental data from published literature^{52,82,84,85}.

D-lactate producer successfully. Finally, we evaluate the economic feasibility of D-lactate production using corncob residue hydrolysate (CRH) through techno-economic analysis (TEA) and life cycle assessment (LCA). We envision that the dominant-metabolism compromised intermediate-chassis will be a valuable strategy for efficiently guiding biochemical production in many microorganisms for circular economy.

Results

Integration of enzyme constraints to improve the genome-scale metabolic model

High-quality GEMs play critical roles in the rational design of microbial cell factories in the classical Design-Build-Test-Learn cycle of synthetic biology studies⁴¹. A high-quality GEM *i*ZM516 of *Z. mobilis* ZM4 was constructed recently with the MEMOTE evaluation score of 91% among all published models, which contains 1389 reactions, 1437 metabolites, 516 genes, and 3 cell compartments⁴¹. Recently, numerous ecModels have been applied to model microorganisms such as *E. coli, S. cerevisiae, C. glutamicum*, and *B. subtilis* with superior potential to identify rate-limiting enzymes, simulate overflow metabolism, and elucidate the trade-off between biomass yield and enzyme usage efficiency^{44,45}.

The previous model of iZM516 built upon the ModelSeed database differs from iZM4_478⁴⁶ in terms of Gene-Protein-Reaction (GPR) relationships by the BiGG database⁴⁷. The unique genes and reactions from *i*ZM4_478 were manually curated and incorporated into *i*ZM516 to obtain the revised iZM547 in this study that contains 2501 reactions, 1455 metabolites, and 547 genes. However, it is important to note that iZM547 only considers stoichiometric constraints and does not reflect the cellular status or provide targets that limit flux through specific product synthesis pathways. With the accumulation of enzyme kinetic data, the predictive accuracy of models can be further enhanced by integrating enzyme constraints that reflect limitations related to protein resources during cell growth. Hence, we applied ECMpy2⁴⁴ and K_{cat} values provided by AutoPACMEN⁴⁸, which is the closest to the experimental results and more accurate to other methods such as DLk_{cat}⁴⁹, TurNup⁵⁰, and UniKP⁵¹, for developing an enzyme-constrained model (ecModel) (Supplementary Fig. 1). The resulting eciZM547_AutoPACMEN mean (subsequently abbreviated as eciZM547) is the enzyme-constrained metabolic network model closest to the experimental results (Supplementary Fig. 1), and ultimately selected for the subsequent analyses.

A relatively comprehensive comparison between *i*ZM516 and *i*ZM4_478 has already been conducted in previous work⁴¹. The results indicate that *i*ZM516 exhibits superior simulation accuracy regarding

strain behavior according to the report⁴¹. As shown in Fig. 2, small discrepancies between *i*ZM516 and ec*i*ZM547 were observed, mainly manifested in the existence of a maximum in the enzyme-constrained model ec*i*ZM547 when the value of glucose uptake exceeds around 71 mmol·gDW⁻¹·h⁻¹. This indicates a shift from glucose-limited growth to proteome-limited growth. The maximum predicted growth rate and ethanol production rate were 0.50 h⁻¹ and 134.76 mmol·gDW⁻¹·h⁻¹, respectively (Fig. 2). Therefore, ec*i*ZM547 exhibited better accuracy than the previous model, which highly overestimated the maximum growth rate and ethanol production rate.

Additionally, iZM516 indicates that most carbon sources are directed towards acetate based on growth criteria when glucose was served as the sole carbon source with an uptake rate of 10 mmol·gDW⁻¹·h⁻¹ under aerobic conditions, whereas eciZM547 calculates that carbon sources flow into both acetate and acetoin, aligning more closely with our previous data³⁶. To further investigate the aforementioned simulation results, we conducted the ¹³C-metabolic flux analysis (MFA) of ZM4 under aerobic condition, and subsequently calculated the distributions of metabolic flux (Supplementary Fig. 2a). Consistent with the predictions made by eciZM547, a greater influx of carbon sources into both acetate and acetoin was observed compared to the anaerobic condition reported before⁵². In addition, in the classical revised model iZM516, the growth rate increased linearly with the substrate uptake. In contrast, the enzyme constraint model of eciZM547 narrowed the solution space, resulting in predictions that are closer to actual experimental conditions in Z. mobilis (Fig. 2).

To investigate the potential of endogenous C2-C5 biochemicals synthesis pathway in *Z. mobilis* ZM4, we calculated and evaluated the metabolic pathways, flux rates, and theoretical conversion rates of biochemicals through simulation and analyses using the *i*ZM516 and ec*i*ZM547 models, respectively (Table 1).

Besides ethanol, lactate, isobutanol, 2,3-BDO, and xylitol also had the same rates under anaerobic or aerobic conditions (Table 1), revealing the advantages of *Z. mobilis* in producing these biochemicals anaerobically, which is crucial for commodity biochemical production since production anaerobically can save the cost around 30–50%. The yield of 1,4-BDO decreases under enzyme constraints (Table 1). However, as the only microorganism using the ED pathway under anaerobic conditions, the conversion rate of 1,4-BDO remained consistent regardless of anaerobic or aerobic conditions as predicted by either *iZ*M516 or ec*iZ*M547 models (Table 1).

Although the rate of succinate production under aerobic condition with the supplementation of CO_2 was higher than that under anaerobic condition, it remains feasible to develop a cell factory of Z.

Table 1 | Metabolic flux rate and theoretical yield under anaerobic and aerobic conditions calculated by iZM516 and eciZM547

		Molecular weight	iZM516_anaerobic ^a		iZM516_aerobic		eciZM547_anaerobic ^b		eciZM547_aerobic	
			mol/mol	g/g	mol/mol	g/g	mol/mol	g/g	mol/mol	g/g
C2	Ethanol ^c	46.00	2.00	0.51	2.00	0.50	2.00	0.51	2.00	0.51
	EG (xylose)	62.00	1.00	0.16	1.00	0.16	1.00	0.16	1.00	0.16
	EG ^d	62.00	0.59	0.09	1.00	0.16	0.22	0.03	0.27	0.04
C3	Lactic acid ^c	89.00	2.00	0.99	2.00	0.99	2.00	0.99	2.00	0.99
	1,3-PDO ^d	76.00	0.50	0.21	0.75	0.32	0.50	0.21	0.74	0.31
	1,3-PDO (glycerol) ^d	76.00	0.75	0.82	0.75	0.82	0.75	0.82	0.75	0.82
	L-alanine ^d	89.00	0.80	0.39	1.33	0.66	0.78	0.38	1.03	0.50
C4	Isobutanol ^c	74.00	1.00	0.41	1.00	0.41	1.00	0.41	1.00	0.41
	1,4-BDO (xylose)°	90.00	0.80	0.40	0.80	0.40	0.77	0.38	0.77	0.38
	Succinic acid ^d	116.00	1.40	0.90	1.71	1.10	1.35	0.87	1.71	1.10
	2,3-BDO ^c	90.00	1.00	0.50	1.00	0.50	1.00	0.50	1.00	0.50
C5	Itaconic acid ^d	128.00	0.38	0.27	1.00	0.71	0.38	0.27	1.00	0.71
	Xylitol ^c	152.00	1.00	0.84	1.00	0.84	1.00	0.84	1.00	0.84

Anaerobic represents model was performed under anaerobic condition in minimal media with oxygen uptake rate set as 0, glucose as the sole carbon source with 40 mmol·gDW⁻¹·h⁻¹ uptake rate, ammonia as the sole nitrogen source with 1000 mmol·gDW⁻¹·h⁻¹ uptake rate.

Aerobic represents oxygen uptake rate set as 1000 mmol·gDW⁻¹·h⁻¹.

Succinate production in aerobic condition (1.71 mol/mol, 1.1 g/g) requires not only oxygen supply but also carbon dioxide supply (1000 mmol·gDW⁻¹·h⁻¹).

EG ethylene glycol, BDO butanediol, PDO propanediol, Except for Ethylene glycol (xylose) and 1,4-BDO (xylose) that use xylose and 1,3-PDO (glycerol) that use glycerol as the carbon source, glucose was used as the carbon source for others.

^aiZM516 represents the published genome-scale metabolic network model.

beciZM547 represents the improved iZM516 with enzyme constraints.

°The rates under anaerobic and aerobic are the same.

^dThe rates under anaerobic and aerobic are different, and the bold font represents the rate is greater than 0.40.

mobilis for anaerobic succinate production at a rate of 0.9 (Table 1). The significant disparity in production rates between anaerobic and aerobic processes for itaconic acid and L-alanine indicates that *Z. mobilis* is more effective for anaerobic production, as suggested by eciZM547 simulation results (Table 1). Moreover, the low rates of ethylene glycol with glucose and xylose as carbon sources under aerobic or even less under anaerobic conditions imply that *Z. mobilis* is not an appropriate chassis for ethylene glycerol production (Table 1). In contrast to ethylene glycol, the rate of 1,3-PDO when utilizing glucose as a carbon source is relatively low. However, a higher rate of 1,3-PDO derived from glycerol suggests its feasibility for production using *Z. mobilis* (Table 1).

Construction of microbial cell factories for C2-C5 biochemical production

A series of recombinant strains were then constructed for biochemical production based on model-guided pathway design, including succinate from glucose, 1,3-PDO from glycerol, xylonic acid, ethylene glycerol, glycolic acid, and 1,4-BDO from xylose (Fig. 3a).

Succinic acid has been designated as one of the top 12 building block chemicals by the U.S. Department of Energy for its unique advantages in chemical, food, agricultural, and plastic industries⁵³, and microbial cell factories using E. coli and other microorganisms have been developed for succinic acid production⁵⁴. Previous model simulations employing iZM516 have indicated that succinic acid production under anaerobic condition using Z. mobilis exhibits significant economic advantages⁴¹. The model-guided metabolic pathway was subsequently optimized through the co-expression of genes involved in the reductive TCA cycle and the individual expression of malate dehydrogenase (encoded by mdh) from C. glutamicum to enhance succinic acid production (Supplementary Fig. 3a). Phosphoenolpyruvate carboxylase (encoded by ppc), fumarate hydratase (encoded by *fumC*), and fumarate reductase (encoded by *frd*) from different sources were screened and tested using the strain SA3 with mdh overexpression as the parental strain (Supplementary Fig. 3b-d). Although there was no significant increase in succinic acid production in these recombinant strains (Fig. 3b), enhanced CO₂ supply can significantly improve succinic acid production, as predicted by the model (Table 1, Fig. 3b). Further overexpression of *mdh* and *ppc* from *C. glutamicum*, *fumC* and *frd* from *S. cerevisiae* in *Z. mobilis* with the supplementation of NaHCO₃ can generate 560 mg/L succinic acid (Fig. 3b). Despite attempts to divert carbon from ethanol to succinic acid using the CRISPRi system, no significant improvement was achieved owing to the robust metabolic characteristic of the ethanol pathway and redox imbalance present in *Z. mobilis* (Supplementary Fig. 4).

Xylose is the second most abundant sugar, accounting for 18-30% of lignocellulose components. Efficient utilization of xylose is crucial for economical biochemical production from renewable biomass⁵⁵. Z. mobilis has been engineered to produce ethanol from xylose by introducing the xylose assimilation and pentose phosphate pathway genes^{56,57}. Xylose dehydrogenase (Xdh) catalyzes the first step of the Dahms pathway from xylose to xylonic acid (Fig. 3a). The codonoptimized xdh gene from Paraburkholderia xenovorans⁵⁸ driven by the strong promoter Ppdc in pXA1 was constructed to produce 16.78 ± 1.58 g/L xylonic acid with CaCO₃ supplementation in RMG2X2 (Fig. 3c). The introduction of the heterologous Dahms pathway could effectively convert xylonic acid into glycolaldehyde and the biosynthesis of ethylene glycol in Z. mobilis based on model prediction using eciZM547. The strong promoter of Peno was used to control the expression of the genes of yagF, yagE, and yqhD from E. coli (Supplementary Fig. 5). Following CaCO₃ supplementation in the recombinant strain EG2 (Fig. 3c), the final ethylene glycerol titer reached 3.26 ± 0.07 g/L. Furthermore, pyruvate generated from the Dahms pathway can be further converted to ethanol (Fig. 3a). By replacing yqhD with aldA, it is possible to achieve a production of 1.5 g/L glycolic acid in the recombinant strain GA (Fig. 3c). In addition, 1,4-butanediol production from xylose can be achieved to 1.2 g/L in the recombinant strain BDO1 by modifying the Dahms pathway and introducing genes including *yagF*, *xylX*, *yqhD*, and *kdcA* (Fig. 3c).

Glycerol is a promising renewable waste feedstock in biotechnology for the production of high-value products such as 1,3-PDO, ethanol, lactate, and succinic acid⁵⁹. 1,3-PDO has various applications in



food, cosmetics, pharmaceuticals, and biomaterials, as it serves as a monomer for the production of polyesters such as polypropylene terephthalate (PTT)⁶⁰. In this study, we tested the feasibility of producing 1,3-PDO using glycerol in *Z. mobilis* by introducing glycerol dehydratase based on eciZM547 model prediction. By comparing different sources of *dhaB* and co-expressing the aldehyde reductase gene *yqhD*, the optimized recombinant strain could achieve 4.1 g/L 1,3-PDO

from glycerol (Fig. 3d). The model prediction results also indicated that the Embden-Meyerhof-Parnas (EMP) pathway has a greater advantage in 1,3-PDO synthesis than the ED pathway of *Z. mobilis* (Supplementary Fig. 6). Therefore, we attempted to complement the truncated EMP pathway of *Z. mobilis* by co-expressing phospho-fructokinase (encoded by *pfk* from *Borrelia burgdorferi*), fructose bisphosphate aldolase (encoded by *fba* from *Z. mobilis*), and

Fig. 3 | **Model-guided pathway design and construction of microbial cell factories for biochemical production using** *Z. mobilis* **as the chassis. a** Modelguided pathway design strategy and metabolic pathway of biochemicals production from glucose, xylose, or glycerol. **b** Summary of succinic acid production in recombinant *Z. mobilis* with different strategies. **c** The titer of ethylene glycerol, xylonic acid, glycolic acid, and 1,4-butanediol in recombinant strains from xylose through Dahms pathway. **d** The titer of 1,3-PDO from glycerol by overexpressing different *dhaB*. The solid represents a one-step reaction, while the dotted line indicates a multi-step reaction. Metabolites: 2D3DXA: 2-dehydro-3-deoxy-D-xylonate, 5HKG: 5-hydroxy-α-ketoglutarate, 6PG: 6-phospho-gluconate, AcCoA: acetyl-CoA, EG: ethylene glycol, GA: glycolate, G3P: glyceraldehyde-3-phosphate, G6P: glucose-6-phosphate, HBA: hydroxy-butyraldehyde, KdcA: keto-acid decarboxylase, KDPG: 2-keto-3-dehydro-6-phosphogluconate, KGAS: ketoglutaric

triosephosphate isomerase (encoded by *tpi* from *Z. mobilis*) to increase glycerol biosynthesis for 1,3-PDO production (Supplementary Fig. 7). The results showed that while introducing the EMP pathway resulted in a slight reduction in glucose consumption and ethanol synthesis of the recombinant strain, the ethanol yield remained unchanged. Notably, however, the glycerol yield increased from 0.28 g/L to 0.8 g/L (Supplementary Fig. 7). Therefore, completion of the EMP pathway proves advantageous for enhancing glycerol production and further utilization of carbon source.

Development of dominant-metabolism compromised intermediate-chassis strategy to reconfigure ethanologen *Z*. *mobilis* into a D-lactate producer

Despite the successful construction of several cell factories in this study aimed at producing C2-C5 biochemicals using glucose, xylose, or glycerol as carbon sources, the titers of these biochemicals remain insufficient for industrialization. This limitation is primarily attributed to the dominant ethanol pathway, which diverts carbon flux to ethanol production. We were unable to directly replace the ethanol pathway with the D-lactate pathway by replacing the *pdc* gene with D-*ldh*. This challenge arises because both pathways utilize pyruvate as the substrate, and are redox and cofactor balanced (Fig. 4a). This result is consistent with a previous report, which indicated that the complete replacement of ethanol with the biosynthesis pathway using inducible promoters to regulate *pdc* expression led to a maximum conversion rate of only 70%⁴³.

The heterologous 2,3-BDO pathway was then introduced first into *Z. mobilis*, which is redox imbalanced with only one NADH comsumption³⁵ (Fig. 4a). Therefore, microaerobic conditions are essential for optimal cell growth to maintain redox balance by oxidizing the extra NADH through respiration with *Ndh*, which also helps maintain the optimal NADH level for cofactor balancing (Fig. 4a). In addition, the growth of *Z. mobilis* is not inhibited by elevated concentrations of 2,3-BDO, and *Z. mobilis* can tolerate to above 100 g/L 2,3-BDO indicating its potential for 2,3-BDO production at high titers³⁵.

Then, the model eciZM547 was used to simulate the result of replacing ethanol pathway with D-lactate pathway in the 2,3-BDO producer background. When glucose was served as the sole carbon source at an uptake rate of 40 mmol·gDW⁻¹·h⁻¹, the quantity of ammonia as the sole nitrogen source was unlimited with biomass as the objective. The results indicated that carbon flux will be shifted toward D-lactate pathway upon knocking out ethanol pathway gene *ZMO1360 (pdc)*. Furthermore, when both ethanol and D-lactate metabolic pathways were knocked out, the enzyme-constrained metabolic model was able to utilize the 2,3-BDO pathway as an alternative for normal cell growth, which is consistent with subsequent experimental results. However, meaningful results cannot be obtained when using eciZM547 to simulate the replacement of ethanol pathway with biochemicals such as isobutanol and 1,3-PDO due to the imbalance in NADH/ NAD⁺ and ATP/ADP ratios.

semialdehyde, OAA: oxaloacetate, PEP: phosphoenolpyruvate, XA: xylonic acid. Enzymes: Adh: alcohol dehydrogenase, AldA: aldehyde dehydrogenase, DhaB: glycerol dehydratase, Frd: fumarate reductase, FumC, fumarate hydratase, Mdh, malate dehydrogenase, Pdc: pyruvate decarboxylase, Ppc, phosphoenolpyruvate carboxylase, Xdh: D-xylose dehydrogenase, XylX: 2-dehydro3-deoxy-D-xylonate dehydratase, YagE: 2-dehydro-3-deoxy-D-xylonate aldolase, YagF: D-xylonate dehydratase, YqhD: aldehyde reductase. Genes: *CbdhaB* from *Clostridium butyricum, CfdhaB* from *Citrobacter freundii, Kppdu* and *kpdhaB* were from *Klebsiella pneumoniae*. Brown font represents carbon source. Blue, green, orange background colors represent C2, C3, C4 compounds, respectively. Data are presented as mean \pm s.e.m. (n = 3 biologically independent samples). Statistical analysis was performed using a two-tailed Student *t*-test. ****p < 0.0001, ***p < 0.001 versus wildtype strain. Source data are provided as a Source Data file.

Hence, the 2,3-BDO producer of Z. mobilis can thus be regarded as the dominant ethanol metabolism-comprised strain under anaerobic condition. This strain was then used in the construction of a D-lactate producer by knocking out pdc gene, using the 2,3-BDO producer as the intermediate chassis (Fig. 4a). As illustrated in Fig. 4b, the wild-type strain ZM4 produces a significant amount of ethanol at a yield of 0.48 g/g. A synthetic operon for high 2,3-BDO production was introduced into the genome of ZM4 to obtain the recombination strain ZMB1, which can produce ethanol, 2,3-BDO, and acetoin at a yield of 0.12, 0.11, and 0.28 (g/g), respectively. To further improve the yield of 2,3-BDO, an additional copy of bdh was integrated at the ZMO0038 chromosomal locus, resulting in the decrease of acetoin and the increase of 2,3-BDO in the recombinant strain ZMB2. Subsequently, the pdc gene was knocked out in ZMB2 to obtain ZMB3 strain, which does not produce ethanol and converts almost all carbon into 2,3-BDO at a yield of 0.42 g/g with only a small amount of acetoin being produced. Therefore, a dominant-metabolism compromised intermediate chassis ZMB3 with 2,3-BDO production was constructed.

To determine the metabolic flux distribution of ZMB3, labeling data from $[1,2^{-13}C]$ tracer experiments with biomass, 2,3-BDO, acetoin production rates were simultaneously fitted to a single flux map of eciZM547 including central carbon metabolism as well as the biosynthesis of 2,3-BDO and acetoin (Supplementary Data 1). During exponential growth, over 96% glucose flux was converted to 2,3-BDO synthetic pathway from ethanol via ED pathway, and the remaining central metabolic reactions, including the TCA cycle and amino acid biosynthesis, had low activity with minimal flux to fulfill biosynthetic needs, which was highly consistent with the results predicted by our model. Additionally, ZMB3 exhibited better fermentation characteristics under aerobic condition compared to ZM4 (Supplementary Table 1), which may be attributed to the enhanced redox balance of ZMB3 under aerobic condition. Therefore, the recombinant strain ZMB3 can be regarded as an excellent aerobic dominant-metabolism compromised intermediate chassis aimed at synthesizing other biochemicals.

Consequently, *bdh* was replaced by *LmldhA* gene encoding D-lactate dehydrogenase to obtain ZMB4 for the production of D-lactate and 2,3-BDO at yields of 0.74 and 0.15 g/g, respectively. To further enhance D-lactate yield, another copy of *LmldhA* was integrated into the chromosomal locus *ZMO1759* to obtain strain ZMB5. Finally, a recombinant strain ZMB6 was constructed with three copies of D-lactate dehydrogenase engineered into 2,3-BDO intermediate chassis (Fig. 4b). To assess the fermentation performance and D-lactate production of ZMB6, shake flask fermentation was carried out using RM medium with varying glucose concentrations. Nearly all glucose was converted into D-lactate with a titer of 140.92 g/L and a yield of 99% by the recombinant strain ZMB6 (Fig. 4c, Supplementary Table 2).

Approximately 40–70% of total costs are allocated to fermentation substrates such as sugars in microbial lactate production⁶¹. The potential of ZMB6 using non-food feedstock of corncob residue



Fig. 4 | **Development of dominant-metabolism compromised intermediatechassis strategy to reconfigure ethanologen** *Z. mobilis* **into a lignocellulosic D-lactate producer. a** Construction and application of the 2,3-BDO intermediatechassis for D-lactate production in *Z. mobilis.* **b** Process of constructing D-lactate producer of *Z. mobilis* using the 2,3-BDO intermediate-chassis. **c** Flask fermentation results of glucose consumption and D-lactate production of the recombinant D-lactate producer ZMB6 in different concentrations of glucose. **d** Fermentor

results of glucose consumption and D-lactate production of the recombinant strain ZMB6 using corncob residue hydrolysate (CRH). Adh: alcohol dehydrogenase, AldC: acetolactate decarboxylase, Als: acetolactate synthase, Bdh: butanediol dehydrogenase, LmldhA (Ldh): lactate dehydrogenase from *L. mesenteroides*, Pdc (encoding by *ZMO1360*): pyruvate decarboxylase. Data are presented as mean \pm s.e.m. (n = 2 biologically independent samples). Source data are provided as a Source Data file.

hydrolysate as a substrate was then evaluated. The results demonstrated a high conversion rate of glucose to D-lactate (Fig. 4d). Due to the advantages of abundance, low price, and ability to mitigate conflicts with food supplies, it is an attractive and promising approach for lactate production using the non-food feedstocks. In a scale-up fermenter with corncob residue hydrolysate, ZMB6 also demonstrated excellent glucose conversion capacity with a D-lactate titer of 104.6 g/L at a yield exceeding 0.97 g/g within 45 h (Fig. 4d), along with an optical purity of 99.1% (Supplementary Tables 2 and 3). These findings indicated the significant advantages in commercial applications.

TEA and LCA of D-lactic acid cell factory

The economic feasibility of utilizing lignocellulosic residues from corncob waste for D-lactate production by recombinant *Z. mobilis* was evaluated (Fig. 5a). The designed production scenario demonstrated the potential to generate over 31,100 tons of D-lactate annually, representing approximately 1.7% of the global lactic acid market, thus establishing a significant market share for the emerging bioeconomy. The comprehensive depiction of assumptions and carbon flow diagram are provided in Supplementary Table 4 and Supplementary Fig. 10. Additionally, the total capital investment (TCI) and total operating cost (TOC) associated with the proposed processes were calculated and listed in Supplementary Tables 5 and 6. Our analysis reveals that with the advantages of lower feedstocks price, higher yield, and scale-up effect in this study, the minimum selling price (MSP)

for D-lactate amounted to USD 0.35 kg⁻¹, remarkably lower than the current market price for lactic acid, set at USD 3.15 kg⁻¹, serving as a benchmark for comparison^{62,63}. When considering the cost of corn stover pretreatment as part of the capital investment (represents approximately 12.89% of the total cost for equipment and installation⁶⁴), the MSP of lactic acid is estimated to be around USD 0.37 kg⁻¹. The competitive advantage of MSP primarily derives from utilizing corncob residues, typically regarded as industrial waste, with a feedstock cost of merely \$38.7 per tonne. This is substantially lower than alternative carbon sources like corn dextrose or molasses (Supplementary Table 7), leading to a significant reduction in raw material costs and TOC. The titer, productivity, and yield of D-lactate of this study are 104.6 g/L, 2.32 g/L/h, and 97%, respectively, which are higher than the values reported in previous studies (Supplementary Table 7). The comparisons of the economic feasibility (TCI, TOC, MSP) of lactate from previous literature reports (Supplementary Tables 7 and 8) reveal that the MSP identified in this study aligns with prior findings and falls within the lower range of reported MSP, which fluctuate based on production capacity, feedstock types, and process design, thereby indicating the economic feasibility of D-lactate production using the recombinant strain ZMB6 developed in this study.

As depicted in Fig. 5b, a single-point sensitivity analysis was conducted to assess seven cost-driving forces based on reasonable ranges, which illustrated the maximum and minimum values of the MSP of D-lactate derived from corncob residues. The analysis demonstrated



Fig. 5 | **TEA and LCA of D-lactate production using the corncob residue. a** Schematic diagram of the industrial production process of D-lactate. **b** Singlepoint sensitivity analysis of the minimum selling price to produce 31,100 tons/year of D-lactate. **c** Minimum selling price (MSP) of lactic acid in various prospective

scenarios. **d** Contribution analysis of the cradle-to-gate global warming potential to produce 1 kg of D-lactate. **e** Global warming potential comparison of producing 1 kg of lactic acid and fossil-based organic acid. D-LA: D-lactate.

that the MSP fluctuated from USD 0.29 to 0.45 kg⁻¹. Importantly, the high sensitivity to annual plant capacity is aligned with the supply of corncob residue, thereby influencing the TCI, TOC, and MSP. Doubling the corncob residue supply from 5000 to 10,000 kg/h leads to a decrease in MSP to USD 0.29 kg⁻¹. In addition, D-lactate productivity and titer stand out as the most critical production factors influencing overall costs. Reduced productivity and titer can result in a notable increase in MSP, largely attributed to the substantial correlation between TCI and annual plant capacity. However, the impact of productivity and titer on MSP decreases above 1g/L/h and 50g/L, respectively (Supplementary Fig. 9), indicating a reduced impact at certain levels of productivity and titer.

A comparison using combined sensitivity analysis can provide further insights into how improving fermentation parameters affects MSP (Fig. 5c). The analysis includes the base case (baseline in the single-point sensitivity analysis), short-term case, and long-term case. Rational predictions for lactic acid productivity, titer, and yield were made based on the optimization of genetic engineering. The selection of long-term factor values was determined by the highest value observed in the single-point sensitivity analysis. The MSP for the short-term and long-term scenario can be USD 0.31 kg⁻¹ and USD 0.28 kg⁻¹, which is reduced by 11% and 20% compared to the base case, respectively. Although the MSP of the base case has shown impressive economic performance, improvements in fermentation-related parameters would further enhance its economic viability.

In addition to economic considerations, a preliminary LCA was also conducted, including results and process simulation using Aspen Plus v14, to demonstrate the global warming potential (GWP) associated with the production of 1 kg D-lactate. The GWP of producing 1 kg D-lactic acid from the corncob residue hydrolysate was determined to be 0.49 kg CO₂-eq (Fig. 5d). A contribution analysis was further performed to identify environmental hotspots throughout the process. The utilization of cellulolytic enzyme (Cellic[®] CTec3, Novozyme, Denmark) was the most significant driver, accounting for 57.5% of all categories, followed by the conversion process at 39.8%. The proposed pathway for D-lactate production has the potential to further reduce greenhouse gas (GHG) emissions by integrating alternative renewable energy sources, such as solar, wind, and geothermal energy.

Upon comparing production data from various non-food feedstocks, including lignocellulosic biomass, vine shoots, corn stover, sugarcane bagasse, and brown leaves, with the corncob residue hydrolysate used in this study, it was noteworthy that the biorefinery GWP demonstrated a reduction of up to 22.51 tons of GHG emissions per ton of D-lactate produced⁶⁵⁻⁶⁸ (Fig. 5e). Moreover, the utilization of this pathway could result in a reduction of GHG emissions by up to 98.16% when compared to fossil-based organic acids (Fig. 5e). Thus, D-lactate production with *Z. mobilis* from corncob residue hydrolysate processes not only offers a practical method for reusing biomass in an industrial setting but also has the potential to mitigate local GHG emissions and ultimately to reduce the overall carbon footprint.

Discussion

The success of circular economy requires the identification and development of efficient chassis cells with diverse industrial characteristics, enabling us to engineer them as biorefinery cell factories for the economic production of a broad spectrum of biochemicals using non-food substrates such as lignocellulosic biomass, molasses, and CO₂^{69,70}. Model microorganisms have been mostly studied and developed due to their clear genetic background with rich knowledge, efficient genome-engineering toolkits, and abundant biological parts and devices. However, model species often lack industrial characteristics such as vulnerability to extreme environmental conditions and toxic inhibitors. In addition, the ease of genome engineering for model species could also lead to proneness of phage infection and genome instability^{71,72}. Therefore, there is growing interest in exploring nonmodel species with excellent industrial features for diverse biochemical production using a wide range of substrates⁷². Taking the recalcitrant polyploid ethnologic bacterium Z. mobilis as an example, we have developed a strategy to help develop non-model species biorefinery chassis.

An improved GEM model of eciZM547 was first constructed in this study by incorporating enzyme constraints, which provides a more

accurate prediction of metabolic pathway than the previous *i*ZM516 to guide rational design of metabolic pathways for creating efficient cell factories. For example, ec*i*ZM547 verified that 1,3-PDO production via the EMP pathway anaerobically in *Z. mobilis* is superior to that under aerobic condition. Guided by the improved GEM, a series of cell factories were successfully constructed using *Z. mobilis* as the chassis. These cell factories are capable of producing various biochemicals, including succinic acid from glucose, ethylene glycol, glycolic acid, and 1,4-butanediol from xylose, and 1,3-PDO from glycerol (Fig. 6).

More importantly, the GEM model simulation also suggested that *Z. mobilis* is well-suited for the economic production of ethanol, lactic acid, isobutanol, 2,3-BDO, and xylitol under anaerobic conditions, providing a direction for bioproducts selection in industrial applications (Table 1). Under aerobic conditions, a more significant yield increase was observed for succinic acid and itaconic acid, two compounds related to the TCA cycle. Therefore, further modifications are necessary to achieve efficient production under aerobic conditions (Table 1). With the development of artificial intelligence (AI)-driven approaches, more accurate and extensive GEMs will be developed to intelligently simulate various metabolic activities within microbial chassis cells to expedite the development of efficient microbial cell factories for industrial applications⁴⁴.

To overcome the challenge of recalcitrant microorganisms with dominant native pathways, we developed a paradigm of dominantmetabolism compromising intermediate-chassis strategy with low toxicity and cofactor relative imbalanced to reconfigure the ethanologenic bacterium *Z. mobilis* into a D-lactate producer. We constructed an intermediate chassis of *Z. mobilis* that compromises its dominant ethanol pathway while producing 2,3-BDO, which was subsequently employed to completely reconfigure the recalcitrant ethanologenic bacterium *Z. mobilis* into a D-lactate producer. The recombinant strain ZMB6 can produce pure D-lactate from glucose and corncob residue hydrolysate with a titer of 140.92 g/L and 104.6 g/L at a yield exceeding 0.97 g/g, respectively. TEA and LCA also confirmed that D-lactate produced from corncob residue hydrolysate can effectively mitigate the environmental impact in industrial applications, and is economically feasible at a price of USD 0.35 kg⁻¹. It is worth noting that although the yield of D-lactic acid increases with prolonged fermentation time, productivity is significantly reduced as a trade-off. Therefore, it is crucial to further improve overall economy viability by increasing product titer, rate, and yield (Supplementary Fig. 9).

In addition to *Z. mobilis, S. cerevisiae* is another model ethanologenic microorganism with excellent industrial merits. However, inactivation of PDCs strains is notoriously known for inability to grow in the presence of excess glucose medium due to NADH/NAD⁺ imbalance under aerobic condition⁷³. Similar to *Z. mobilis*, recent research has determined that the introduction of the 2,3-BDO pathway directly into *S. cerevisiae* results in a coupled anaerobic 2,3-BDO and glycerol production through the ecGEM prediction in ethanol block-out strains⁷⁴. A previous study also showed that the introduction of 2,3-BDO and lactic acid pathway significantly improved NADH-consuming⁷³. Hence, we propose that a dominant-metabolism compromising intermediatechassis strategy remains applicable to *S. cerevisiae* for anaerobic biochemicals production, and a more comprehensive GEM in *S. cerevisiae* can efficiently guide biochemicals production without ethanol constraints.

The unique physiological and genetic features of *Z. mobilis* evolved to adapt to the unique environment, make it a robust chassis for pyruvate-derived extracellular biochemical production such as the innate ethanol biosynthesis pathway or the D-lactate production pathway constructed in this study. The successful reconfiguration of the ethanologenic bacterium *Z. mobilis* into several cell factories for C2-C5 biochemical production using glucose, xylose, or glycerol as substrates in this study (Fig. 6) demonstrates its feasibility to develop *Z. mobilis* as efficient cell factories for the commercialization of other biochemicals such as L-alanine, isobutanol, and 2,3-BDO using pyruvate as the direct precursor^{32,25,36}.

However, it is disadvantageous to construct cell factories for producing energy-consuming biochemicals using *Z. mobilis* since there is only one molar of ATP produced from one molar of glucose because of its unique anaerobic ED pathway and truncated EMP and TCA. For example, compared to the classical EMP pathway, the only known anaerobic ED pathway provides efficient substrate utilization but limits its ability to synthesize compounds such as L-serine and



Fig. 6 | Summary of biochemicals production in Z. mobilis. 2-KIV 2-Ketoisovalerate, ED Entner-Doudoroff pathway, PEP phosphoenolpyruvate, PPP pentose phosphate pathway, TCA tricarboxylic acid cycle. Purple font represents substrate. Black font indicates the relevant products in this work.

1,3-PDO³³ using glyceraldehyde-3-phosphate as precursor. Additionally, incomplete TCA results in high energy utilization efficiency with declined biomass, which hinders the economic production of intracellular biochemicals³⁹.

Although metabolic engineering, random mutagenesis, and adaptive laboratory evolution can be applied to reconfigure *Z. mobilis* for enhanced energy production and biomass generation, designing and predicting these modifications remains a significant challenge. This difficulty arises from a significant knowledge gap in gene functions, physiological and information metabolism, as well as regulatory signals and networks. Therefore, more microbial chassis with different advantages are needed, and it is crucial to understand the strengths and weaknesses of both model and non-model microorganisms when considering them for biotechnology applications. Additionally, it is crucial to leverage their inherent advantages while introducing metabolic or regulatory pathways in the design of biorefinery cell factories.

In addition, it is also important to integrate AI technology and big data analysis with microbial physiological, metabolic, and omics datasets of diverse biochemical producers under different conditions to construct high-accuracy mathematical models⁷⁵. This will help predict and enhance the adaptability of metabolic pathways to specific chassis cells, improve product biosynthesis ability, and enhance strain robustness. With the rapid development of mass spectrometry, highresolution microscopy, as well as efficient gene sequencing, editing, and synthesis techniques, more microorganisms with unique industrial features will be identified and characterized, and the corresponding genome information and genome-engineering toolkits can then be developed rapidly to help engineer them as biorefinery cell factories. Furthermore, the accumulation of physiological and omics datasets along with an increased understanding of the underlying mechanisms of cell organization, regulation, and proliferation, the increased knowledge of biological parts, devices and systems, and effective systems and synthetic biology approaches will facilitate rational design, synthesis assembly, and delivery of artificial genomes and synthetic organisms as the ultimate universal chassis for circular bioeconomy development.

Methods

Strains and media

Z. mobilis ZM4 was employed for the construction of biochemicals production strain. *E. coli* DH5 α and Trans110 were used as a base strain for plasmid construction and demethylation. Plasmids, strains and oligonucleotides used in this study are listed in Supplementary Data 2.

E. coli was cultured at 37 °C, 250 rpm using Luria-Bertani (LB) medium (10 g/L tryptone, 5 g/L yeast extract, 10 g/L NaCl), and Rich medium (RMG5, 50 g/L glucose, 10 g/L yeast extract, 2 g/L KH₂PO₄) was used for *Z. mobilis* cultivation at 30 °C, 100 rpm. RMG2X2 (20 g/L glucose, 20 g/L xylose, 10 g/L yeast extract, 2 g/L KH₂PO₄) was used for cultivation of the xylose utilization recombinant strains. 100 μ g/mL spectinomycin was added for both *E. coli* and *Z. mobilis* when appropriate.

Genetic manipulation

The acquisition of fragments and linear vectors, as well as the plasmids construction process, were carried out according to previously reported method⁷⁶. Specifically, the insert(s) and vector were ligated through T5 exonuclease kit (NEB, USA) with a molar ratio of 3:1. The ligation product was used for transformation into *E. coli* Dh5 α competent cells. The transformants were confirmed by colony PCR using specific primers of 15A-Fwd/Rev to confirm the insert size, correct plasmids were further verified by Sanger sequencing (TsingKe, Wuhan, China). Genetic modification of *Z. mobilis* was based on native Type I-F CRISPR-Cas system by editing plasmid pL2R²⁷. For gene insertions, the gene expression cassette flanking 600 bp upstream and 600 bp downstream of editing sites was assembled into the pL2R linearized vector. The correct plasmid was then transferred into *Z. mobilis*

competent cells via electroporation⁷⁷. The CRISPRi system application for inhibiting *pdc* and *ldh* expression was then carried out³⁴. The targeting gRNA sequence was annealed using two single-stranded oligonucleotides and ligated into linearized pEZ-sgr which carries a minimal CRISPR array.

C2-C5 biochemicals metabolic pathway construction

Heterologous gene sequences using for C2-C5 biochemicals production in this study were listed in Supplementary Data 3, these genes were synthesized from TsingKe (Wuhan, China) after codonoptimized. The promoter used were cloned from Z. mobilis ZM4 genome. For constructing the succinate synthesis pathway in Z. mobilis. The strong promoter Peno was used to drive mdh (encoding malate dehydrogenase) genes from different sources on plasmid to evaluate the efficiency (Supplementary Fig. 3a). The expression cassette with the higher titer was integrated into the ZMO0028 locus of Z. mobilis genome to obtain SA3 strain. Based on SA3, different ppc (encoding phosphoenolpyruvate carboxylase) genes driven by Peno were tested in Z. mobilis (Supplementary Fig. 3b). The better efficiency of ppc from C. glutamicum was inserted into the genome of SA2 at ZMO0038 locus to obtain SA3 strain. Similarly, the other genes fum (encoding fumarate hydratase) and frd (fumarate reductase) cloned from different sources were overexpressed on plasmid and transformed to SA3 for testing (Supplementary Fig. 3d, e). Then the efficient synthetic operon of fumfrd driven by Ptet was constructed and transformed into SA3 for succinic acid production.

Similarly, a synthetic operon was constructed to overexpress different source of genes dhaB (encoding glycerol dehydratase) driven by strong promoter Ppdc, and gdrAB (encoding activating factor), yqhD (encoding aldehyde reductase) driven by strong promoter Peno on plasmid for 1,3-PDO production. The sequence of RBS used for gdrAB and yqhD connecting is 5' ATTAAAGAGGAGAAA 3'. For xylonic acid production, Ppdc promoter was used for xdh overexpressed to obtain plasmid pXA1, ethylene glycol, glycolate, and 1.4-butanediol production, the genes yagF (encoding xylonate dehydratase), yagE (encoding 2-keto-3-deoxy D-xylonate aldolase), and yahD (encoding aldehyde reductase) were cloned from E. coli. These genes were overexpressed by Peno promoter and inserted into pXA1 plasmid to obtain pEG2 for ethylene glycol production. Gene yqhD of pEG2 was replaced with aldA to obtain pGA1 plasmid for glycolate production. For achieving 1,4-BDO production from xylose in Z. mobilis, the gene yagF of pEG2 was replaced with xylX to obtain pBDO1 plasmid. Then the pBDO1 plasmid was transformed into ZMQ3 strain³⁶ for 1,4-BDO production. The gene LmldhA driven by constitutive promoter PadhB (encoding D-lactate dehydrogenase) for lactic acid production was chosen and used in this work^{31,36}.

Electroporation transformation and recombinant strain selection

The plasmids were transformed into electrocompetent cells of *Z. mobilis* by electroporation (Bio-Rad Gene Pulser, 0.1-cm gap cuvettes, 1.6 kV, 200 ohms, 25 μ F). The electrocompetent cells were transferred and recovered in RMG5 medium at 30 °C for 3 h³⁶. For recombinant strain selection, transformants were cultured on RM agar plate containing 100 μ g/mL spectinomycin (RMSp). Potential mutants were screened by colony PCR with the specific primer pair. Then the current recombinant strains were further cultivated in RMSp medium for preservation and further flask evaluation. For curing of editing plasmid, transformants with correct PCR results were cultivated in RM media without antibiotics at 30 °C and passaged for five generations.

Construction of dominant-metabolism compromised intermediate chassis

Z. mobilis ZM4 was used as the initial strain and conducted three rounds of genome editing (Fig. 5b). First, we replaced the *ZMO1650* of

ZM4 with 2,3-BDO expression cassette including *Bsals* driven by inducible promoter *Ptet* and a synthetic operon *Aldc-Bdh* driven by *Pgap* promoter, resulting in ZMB1, which enabled the heterologous expression of the 2,3-BDO pathway. Then, the *ZMO0038* of ZMB1 was replaced by *bdh* driven by *Pgap promoter*, generating ZMB2 for further optimizing the metabolic flux of the 2,3-BDO pathway. Finally, the *ZMO1360* (*pdc*) of ZMB2 was deleted by editing plasmid pB3 to obtain dominant-metabolism compromised intermediate-chassis ZMB3, completely redirecting the central carbon flow from the ethanol pathway to the 2,3-BDO pathway.

In order to further redirect carbon flux towards D-lactic acid production, we replaced the *ZMOOO38-bdh* gene of ZMB3 with *Dldh*, resulting in ZMB4, which achieved the heterologous expression of the D-lactic acid pathway and redirected carbon flow from the 2,3-BDO pathway to the D-lactic acid pathway. Subsequently, we replaced the *ZMO1759* gene of ZMB4 with *Dldh* to obtain ZMB5, which increased the copy number of lactate dehydrogenase and further competed for metabolic carbon flow, consequently enhancing the flux of the D-lactic acid metabolic pathway. Finally, we replaced the *ZMO1650*-2,3-BDO expression cassette of ZMB5 with *Dldh* to obtain ZMB6.

Batch fermentation in shaking flasks

Recombinant strains were pre-cultured overnight with OD_{600nm} value of 1.0–2.0 in RMG5 medium for the seed culture preparation, which was inoculated in 50 mL shake flasks containing 40 mL corresponding medium with an initial OD_{600 nm} value of 0.1 after being collected and washed twice. During fermentation, cell growth in terms of the absorbance value (OD_{600 nm}) was determined spectrophotometrically at 600 nm (UV-1800, AOE, China) at different time points. Samples were centrifuged at 17,000 × g for 2 min, collected supernatants were passed through 0.22- μ m filters and stored at –80 °C for subsequent HPLC analysis if needed.

For biochemicals production from xylose, the seed culture was transferred into 50 mL shake flasks containing 40 mL RMG2X2 medium with an initial OD_{600nm} value of 0.1 at 30 °C, 100 rpm. 10 g/L CaCO₃ was supplemented to control the pH during fermentation. Samples from different time periods were collected and stored at -80 °C for subsequent HPLC analysis.

For 1,3-PDO production from glycerol, the seed culture was transferred into 50 mL shake flasks containing 40 mL RMG5 medium with an initial OD_{600nm} value of 0.1 at 30 °C on a rotary shaker at 100 rpm. 10 g/L glycerol and 5 mg/L vitamin B_{12} was supplemented into the media.

For D-lactic acid production from CRH in shake flasks, the seed culture was transferred into medium with an OD_{600nm} value of 0.5 at 33 °C on a rotary shaker at 100 rpm. 2 g/L yeast extract was supplemented into the media.

Production of D-lactate from corncob residue hydrolysate

Corncob residue hydrolysate with approximately 107 g/L glucose (pH 4.50) used in this study was provided by Wuhan Ruijiakang Biotechnology Co., Ltd. (Hubei, China). As for batch fermentation in 5-L fermentor, 50 g/L CaCO₃ and 2 g/L yeast extract was supplemented into medium. Strain ZMB6 at an initial OD_{600nm} value of 0.5 was cultured at 33 °C, 100 rpm for 72 h for the measurements of glucose and D-lactate.

¹³C metabolic flux analysis

 13 C metabolic flux analysis (MFA) of amino acid in the wild-type ZM4 and ZMB3 strains was carried out according to previously reported method^{78,79}. In particular, [1,2–¹³C] glucose (99.1 atom% ¹³C) purchased from Sigma-Aldrich (St. Louis, MO) was used as the sole carbon source for cell growth. Briefly, Cells cultured in MM medium (20 g/L glucose, 1g/L K₂HPO₄, 1g/L KH₂PO₄, 1g/L (NH4)₂SO₄, 0.5 g/L NaCl, 0.42 g/L MgCl₂·6 H₂O, 0.001 g/L calcium pantothenate) were collected at the

exponential phase and hydrolysate by 6 M HCl at 100 °C for 18 h. The supernatant of hydrolysate was freeze-dried by low temperature concentration cold trap. Dried exudates were subsequently dissolved with 100 uL tetrahydrofuran followed by the addition of 100 uL N-tere-Butyldimethylsilyl-N-methyltrifluoroacetamide (TBDMS) at 70 °C for 1 h to generate derivatives. After centrifugation for 5 min. 1-uL derived sample was quantified by GC-MS (Gas chromatograph Trace, Thermo; Triple Quadrupole Mass Spectrometer), fitted with a TG-5MS column (30 mm \times 0.25 mm \times 0.25 µm, Thermo; USA) with 1.2 mL/min flow rate of carried gas. The detector procedure of GC was set as follows: the initial temperature was held at 150 °C for 2 min and was programmed to raise to 280 °C at 3 °C per min, and further increase to 300 °C at 20 °C per min and hold for another 5 min. Solvent delay was set as 5 min, and the range of mass to charge ratio (m/z) in MS was set from 60 to 50078. The software WuFlux-Ms Tool was applied to analyze and correct amino acid MS data (fragments of [M-57]⁺, [M-159]⁺, [M-85]⁺, and [f302]).

¹³C-MFA was calculated by using the INCA software, which is based on the elementary metabolic unit framework. A metabolic reaction network of Z. mobilis central metabolism, including ED and EMP glycolytic reactions, non-oxidative PPP reactions, TCA cycle reaction, lumped amino acid biosynthesis, was constructed from model eciZM547. Data of labeling amino acids were substituted into the model for calculating metabolic flux of central metabolism. Several reactions about acetate, acetoin and 2,3-BDO syntheses including Acetaldehyde + NAD⁺ \rightarrow Acetate + NADH, 2 Acetaldehyde \rightarrow Acetoin, Pyruvate \rightarrow acetolactate + CO₂, Acetolactate \rightarrow Acetoin + CO₂, Acetoin + NADH \leftrightarrow 2,3-BDO + NAD⁺, Acetoin + NAD⁺ \rightarrow Acetyl-CoA + NADH, Acetoin + ATP \rightarrow Acetoin. ext, 2,3-BDO \rightarrow 2,3-BDO. ext were added to the metabolic network model. Cell growth rate, glucose consumption rate and fermentation production rate were characterized for exponential phase growth at 30 °C in MM medium under 10% medium volume condition for using as MFA constraints (Supplementary Table 1). The measured data of labeling amino acid was substituted into the model for calculating metabolic flux (Supplementary Data 1). OD₆₀₀ measurements were converted to dry weight⁵². The flux estimations were performed by using Metlab R2012b (The Mathworks Inc.)

Analytical methods

The concentration of glucose and ethanol in the supernatant was analyzed by HPLC LC-20AD (Shimadzu, Japan) with a RezexTM, RFQ-Fast Acid H⁺ (8%) (Phenomenex, CA, USA) at 80 °C. Elution was performed with 5 mM H₂SO₄ at 0.6 mL/min³⁶. Lactic acid, succinic acid, 1,3-PDO, 1,4-BDO, glycerol, xylose and xylonic acid concentration were determined by HPLC equipped with a Bio-Rad Aminex HPX-87H column (Bio-Rad, Hercules, CA, U.S.A.) and a refractive index detector^{54,60,80}. The column temperature was set at 60 °C and 5 mM H₂SO₄ solution was used as the mobile phase with a flow rate of 0.5 mL/min.

Since xylonic acid and xylose have similar retention times with refractive index detector, xylonic acid has determined using the UV detector when xylose was present by HPLC-2030 Plus (Shimadzu, Kyoto, Japan) equipped with a Bio-Rad Aminex HPX-87H column (Bio-Rad, Hercules, CA, U.S.A.) at 60 °C. 5 mM H₂SO₄ solution was used as the mobile phase with a flow rate of 0.5 mL/min. Xylose concentrations were estimated by subtraction of the xylonic acid peak from the combined peak of xylose and xylonic acid when xylonic acid was present⁵⁸.

Construction of enzyme-constrained metabolic network model The metabolic model of this study is based on *i*ZM516. By comparing with *i*ZM4_478, this study added the unique genes and related reactions of *i*ZM4_478 relative to *i*ZM516, tentatively designated as *i*ZM547, and the detailed changes were included Supplementary Data 4 and 5. The construction of the enzyme-constrained model is mainly completed using the process ECMpy2.

Data collection, molecular weight (MW) and Subnit number data were collected from the Interaction information in the UniProt database based on Gene ID. Four methods (AutoPACMEN, DLk_{cat}, TurNup, and UniKP) were used to determine K_{cat} values⁴⁹⁻⁵¹. AutoP-ACMEN was mainly determined based on the K_{cat} values of various species in the Brenda and SABIO-RK databases using EC numbers. Data of Z. mobilis were directly accessed, otherwise the average or median K_{cat} values of all species under the selected EC number will be used as the K_{cat} value for the relevant reaction. If there is no K_{cat} value, the default K_{cat} value was filled based on the average or median K_{cat} values of all different EC numbers. The protein abundance of Z. mobilis was obtained by selecting PXD030417 protein data from the PRIDE database, and the original data was completed through the online service of MaxQuant and PAXdb⁸¹. The protein-to-cell ratio ptot is set to 0.605^{38,82}. The approximated average saturation of enzyme σ was set as 0.5 in default. Enzyme mass fraction f was calculated according to Eq. 1:

$$f = \frac{\sum_{i=1}^{p-num} A_i M W_i}{\sum_{i=1}^{g-num} A_i M W_i}$$
(1)

where A_i and A_j represented the abundance of the i-th protein (p_num represents proteins expressed in the model) and j-th protein (g_num represents proteins expressed in the whole proteome).

First, reversible reactions in *i*ZM547 were divided into pairs of irreversible reactions, and reactions catalyzed by multiple isoenzymes were split into different reactions (append num in reaction ID, e.g., rxn00001_reverse_num1). Next, the MW of each enzyme was calculated. According to Eq. 2, the total sum of proteins in the complex was used for reactions catalyzed by enzyme complexes. Finally, an enzymatic constraint (Eq. 3) was introduced into the model with function trans_model2enz_json_model_split_isoenzyme. Then, the biomass calculation was calibrated in enzyme-constrained model by identifying the reaction with the enzyme cost and substituting its K_{cat} values with the highest value found in the Brenda or SABIO-RK database.

$$MW = \sum_{j=1}^{m} N_j * MW_j \tag{2}$$

where m is the number of different subunits in the enzyme complex and N_i is the number of j-th subunits in the complex.

$$\sum_{i=1}^{n} \frac{v_i * MW_i}{\sigma_i * k_{cat,i}} \le ptot * f$$
(3)

Process simulation

Schematics of the D-lactate production from corncob residue hydrolysate process is presented in Supplementary Fig. 8. In this study, lactic acid production process was designed using Aspen Plus[®] software, which was also used to calculate mass and energy balance. The D-lactate production process includes five areas: glucose production (A100), D-lactate production (A200), D-lactate purification (A300), wastewater treatment (A400), and utilities (A500). Detailed process descriptions can be found in the Supplementary Method 1. The major assumptions and details of the process design in Aspen Plus v14 can be found in Supplementary Table 4. The mass and energy data collected from Aspen Plus v14 are presented in Supplementary Tables 9–13 and Supplementary Data 6.

Techno-economic analysis

TEA is an integrated process that can be used to assess the feasibility of lactic acid production from corncob residue based on an economic

and technical point of view. The TCI, TOC, and MSP were estimated using an in-house model developed by the National Renewable Energy Laboratory (NREL)⁸³. The mass balance of the lactic acid production process conducted using Aspen Plus v14 is presented in Supplementary Fig. 10. An annual lactic acid production capacity of 31,100 tons for the biorefinery was assumed along with influencing factors such as equipment size, raw material expenditures, and other associated costs. The TEA is an "nth-plant" model that can estimate the investment necessary for the pre-commercial process as well⁸³. The key assumptions and basic parameters for the process scale-up are shown in Supplementary Table 14, which are sourced from NREL's reports and studies⁸³.

The equipment and raw materials costs used in the TCI and TOC estimation were obtained from published literature and official reports (Supplementary Table 15)⁸³. The costs of equipment necessarily change with implementation size and obey a power law of the original prices⁶⁵. The inside battery limit (ISBL) is commonly used in TEA for basic related to warehouse investments, site development, and additional piping. In this study, the ISBL includes equipment and installation costs for A100 to A300. The equipment and raw materials costs will be processed in an Excel spreadsheet using established formulas to calculate the TCI and TOC of the proposed processes (Supplementary Tables 5, 6). Using the discounted cash flow method, the MSP of lactic acid can be determined at the point where the net present value (NPV) equals zero and the internal rate of return (IRR) reaches 10%83. All costs are converted to 2023 US dollars to eliminate the influence of the rate of inflation on expenditure according to the Plant Cost Index from Chemical Engineering Magazine, the Industrial Inorganic Chemical Index from SRI Consulting, and the US Department of Labor Bureau of Labor Statistics⁸³.

Life cycle assessment

The main goal of the LCA was to measure the carbon footprint of manufacturing lactic acid using corncob residue. The system boundary is "from cradle to gate", raw material consumption and production processes were considered. In LCA investigations, the functional unit for chemicals is commonly defined as the mass of the product in kilogram. In order to compare the results to other similar research, this study selected 1 kg of lactic acid as the functional unit to serve as a benchmark for the inputs and outputs. The lactic acid production pathway in this study obtained crucial raw materials and energy inputs from Aspen Plus v14. Additionally, the disposal capability for corncob residue was determined to be 5000 kg/h. Supplementary Table 16 presents the data of materials and energy used in the process of producing lactic acid. The Environmental Protection Agency (EPA) produced version 2.1 of the tool for reduction and assessment of chemicals and other environmental impacts (TRACI), which provides characterization parameters that estimate the potential effects of inputs and discharges on certain manufacturing processes. This study evaluated the GWP of the lactic acid production process. The background data utilized in the life cycle assessment is based on Ecoinvent v3.6.

Statistical analysis

The number of biological replicates is specified in figure and table legends. Data analysis was performed using GraphPad Prism statistical software (version 8.0.1, GraphPad, CA, USA). Independent unpaired *t*-test was used for two-tailed comparison. $p \le 0.05$ was considered statistically significant.

Data availability

Data supporting the findings of this work are available within the paper and its Supplementary Information files. A reporting summary for this article is available as a Supplementary Information file. Source data are provided with this paper.

Code availability

The code was implemented in Python 3.9, and the code relies on the COBRApy and ECMpy packages. Full code and models of eciZM547 are available on GitHub [https://github.com/AaroncrowAries/eciZM547].

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Author contributions

S.Y. designed and supervised the experiments with inputs from all authors. X.Y. and W.B. constructed plasmids and recombinant strains with the help from Z.H., Q.P., M.H., and B.G. Y.W. performed the model construction and optimization with help from Z.M., Q.Y., Q.F., and H.M. X.Y., P.H. performed the ¹³C metabolic flux analysis with the help from S.C. C.Z. and Q.F. performed the techno-economic analysis and life cycle assessment of D-lactate production from corncob residue. X.Y., W.B., Y.W., Z.H., Q.H., Q.F., and S.Y. analyzed the data. X.Y., Q.H., and S.Y. wrote the manuscript with the help from Q.F. and H.M. All authors helped revise, read, and approve the final manuscript.

Competing interests

The authors declare that they have patent applications associated with this study. S.Y., X.Y., Q.H. have filed two patents (CN202310246847.7 and US18401511) for protecting strains for EG production. S.Y., W.B., Q.P. have filed two patents (CN202410112699.4 and US18777118) for protecting strains for D-lactate production. Other authors claim no competing interests.

Additional information

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